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Balancing Chemical Equations - Answer Key Balance The Equations Below: 1) $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$ 2) $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$ 3) $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$ 4) $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$ 5) $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$ 1th, 2024

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$2\text{HCl} \rightarrow \text{H}_2 + \text{Cl}_2$... 6th, 2024

Balancing Chemical Equations - Answer Key

Balancing Chemical Equations - Answer Key Balance The Equations Below: 1) $1 \text{N}_2 + 3 \text{H}_2 \rightarrow 2 \text{NH}_3$ 2) $2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$ 3) $2 \text{NaCl} + 1 \text{F}_2 \rightarrow 2 \text{NaF} + 1 \text{Cl}_2$ 4) $2 \text{H}_2 + 1 \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$ 5) $1 \text{Pb}(\text{OH})_2 + 2 \text{HCl} \rightarrow 2 \text{H}_2\text{O} + 1 \text{PbCl}_2$ 6) $2 \text{AlBr}_3 + 3 \text{K}_2\text{SO}_4 \rightarrow 6 \text{KBr} + 1 \text{Al}_2(\text{SO}_4)_3$ 7) $1 \text{CH}_4 + 2 \text{O}_2 \rightarrow 1 \text{CO}_2 + 2 \text{H}_2\text{O}$ 4th, 2024

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