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CH2=CH2

B - Polarity Like Alkenes, Alkene Are Relatively Nonpolar. They Are Insoluble In Water But Soluble In Nonpolar Solvents Such As Hexane, Gasoline, Halogenated Solvents, And Ethers. A Alkene Tend To Be Slightly More Polar Than Alkanes, However, Because The More Weakly Held Electrons In The Pi Bond Are More Polarizable. Jan 12th, 2024

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