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C2H50H(1) +3 2 C02(g) +3 H20(g) Cot 31) A 100.0 MI. Sample Of 0.300 M NaOH Is Mixed With A 100.0 MI. Sample Of 0.305 M HN03 31) In A Coffee Cup Calorimeter. If Both Solutions Were Initially At 35.0 Oc And The Temperature Of The Resulting Solution Was Recorded As 37.0 Cc, Determine The AHrxn (in Units Of KJ/mol). 23th, 2024

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Unit 06 Thermochemistry And Nuclear Chemistry Notes (answers)

11.3: Heat In Changes Of State Enthalpy (H): - The Amount Of Internal Energy At A Specific Pressure And Volume. Energy Involved In Physical Change (Temperature Or Phase Change): Heating Curve: - A Graph Of Temeparture Versus Time As A Substance Is Heated From A Solid Phase To A Gaseous Phase. 9th, 2024

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