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Should Be. A. $\text{C}_4\text{H}_{10} / \text{O}_2$ B. O_2 / CO_2 C. $\text{O}_2 / \text{H}_2\text{O}$
D. $\text{C}_4\text{H}_{10} / \text{CO}_2$ E. $\text{C}_4\text{H}_{10} / \text{H}_2\text{O}$ 2. Given The
Following Equation: $2\text{KClO}_3 \rightarrow 2\text{KCl}$ 10th,

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For Dummies Stoichiometry Worksheet #1 Answers 1.
Given The Following Equation: $2\text{C}_4\text{H}_{10} + 13\text{O}_2 \rightarrow 8\text{CO}_2 + 10\text{H}_2\text{O}$, Show What The Following Molar
Ratios Should Be. A. $\text{C}_4\text{H}_{10} / \text{O}_2$ B. O_2 / CO_2 C. $\text{O}_2 / \text{H}_2\text{O}$
D. $\text{C}_4\text{H}_{10} / \text{CO}_2$ E. $\text{C}_4\text{H}_{10} / \text{H}_2\text{O}$ 2. Given
The Following Equation: $2\text{KClO}_3 \rightarrow 2\text{KCl}$ 17th,
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