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To 1.56x10²¹ Atoms Of Sodium? 1.56 x 10²¹ Atoms Na X 1 Mol Na = 2.59 X 10³ Mol Na 236.022 X 10 Atoms Na 2) Determine The Mass In Grams Of Each Of The Following: A. 1.35 Mol Of Fe 1.35 Mol Fe X 55.845 G Fe = 75.4 G Fe 1 Mol Fe B. 24.5 Mol O Feb 4th, 2024.

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Practice MC Unit A Atoms, Moles, Stoich And More 4 • ? H 2O Weighing 24.95 G Is Heated Until All The Water Is Removed. The Anhydrate Weighs 15.95 G. What Is The Value Of ? In The Hydrated Salt? (MM CuSO 4, 159.5 G/mol) A. 2 B. 3 C. 4 D. 5 27. The Simplest Formula For A Hydrocarbon That Is 20.0 Percent Hydrogen By M Feb 2th, 2024 Chapter 8 Practice Worksheet: Formulas, Equations, And Moles Percent Composition And Empirical Formulas 8) What Is The Mass Percent Of Each Element In The Following Compounds? A. CaCl 2 B. Fe 2 O 3 C. C 6 H 10 S 2 O 9) Calculate The Empirical Formulas Of Compounds Containing The Following Percentages Of Elements. Use The Molar Mass To Calc Mar 4th, 2024 Practice Packet Unit 7: Moles & Stoichiometry PRACTICE PACKET: Unit 7 Moles & Stoichiometry 5 Www.chempride.weebly.com ADDITIONAL PRACTICE LESSON 1: Find The Gram Formula Mass Of The Following: (Show All Work) 1. MgO 5. Ca(OH) 2 2. NaHCO 3 6. CH 4 3. C 6 H 12 O 6 7. NH 3 4. Al Apr 8th, 2024.

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