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What Would Be The Weight Of This Maximum Yield? To Answer This, We Need To Convert Our Mole Value For H2 S To A Mass Value. Remember That M = NMw We Can Specify $Mw = (2 \times 1,008) + 32.06 = 34.08$ G / Mol Proto $M = 3.5 \times 34.08 = 119.3$ Grams H 2 S Could Be Produced. III. If You Would Like To Produce 3.00 Kg H 2 Apr 4th, 2024

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