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Problems)Stoichiometry Worksheet #2 (mole-mass, Mass-mole Problems) 1. N₂ + 2O₂ → N₂O₄ A. If 15.0g Of N₂O₄ Was Produced, How Many Moles Of O₂ Were Required? 15.0g N₂O₄ 1 Mol N₂O₄ 2 Mol O₂ 92.0g N₂O₄ 1 Mol N₂O₄ = 0.326 Mol O₂ B. If 4.0x10⁻³ Moles Of Oxygen Reacted, How Many Grams Of N₂ Were Needed? 4.0x10⁻³ Mol O₂ 1 Mol N₂ 28 ... Mar 2th, 2024CHEMISTRY WORKSHEET # 2 MOLE PROBLEMS—THE MOLE ...CHEMISTRY WORKSHEET # 2:

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Worksheet: Mixed Problems—Mole/Mole Name And ...2
___ CuO A. If 101 Grams Of Copper Is Used, How Many
Moles Of Copper (II) Oxide Will Be Formed? B. If 5.25
Moles Of Copper Are Used, How Many Moles Of Oxygen
Must Also Be Used? C. If 78.2 Grams Of Oxygen React
With Copper, How Many Moles Of Copper (II) Oxide Will
Be Produced? 2. ___ C 4H 10 + ___ O 2 ___ CO 2 + ___ H
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