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Sheet 1 Directions: Solve Each Of The Following Problems. Show Your Work, Including Proper Units, To Earn Full Credit. 1. Silver And Nitric Acid React According To The Following Balanced Equation: $3 \text{Ag}(s) + 4 \text{HNO}_3(aq) \rightarrow 3 \text{AgNO}_3(aq) + 2 \text{H}_2\text{O}(l) + \text{NO}$... May 13th, 2024.

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Chapter 112 Subchapter C Texas Education Agency April 30th, 2018 - Education Rules Concerning Texas Essential Knowledge And Skills Curriculum Standards For High School Science"
[https Classroom Google Com](https://classroom.google.com) May 3rd, 2018 - "Chemistry With Lab 2018 - Easy Peasy All In One High School May 3rd, 2018 - Please Not Jan 16th, 2024.

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33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$
Actual yield = 1.45g Theoretical yield = 5.95g
 $\text{KClO}_4 \times 1 \text{ mol} \times 122.55 \text{ g} \times \frac{2}{2} = 1 \text{ KClO}_4 \times 32.00 \text{ g} \times 1 \text{ mol} = 3.11 \text{ g O}_2$
%Yield = $\frac{1.45 \text{ g}}{3.11 \text{ g}} \times 100\% = 46.6\%$
34. $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$
Percent purity = 62.8%
Apr 5th, 2024
Practice Work 53 Stoichiometry-04 Mixed Stoichiometry ...
Practice Work 53 - Stoichiometry-04 Mixed Stoichiometry Problems
General Information You Will Need A Periodic Table, Your Stoichiometry Notes, And Appendix 12 For This Assignment.
Sorry About The Lack Of Format. I'm In A Time Crunch.
123.88 G/mol 70.90 G/mol 137.32 ...
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