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Elements Symbol. Element Atomic # Atomic Mass Protons Neutrons Electrons Valance Electrons Number Of Energy Levels Bohr Model Lewis Dot Carbon 6 12 6 6 6 4 (group 14, 4 In 4th, 2024

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Create A Lewis Dot Structure For An Atom And A Molecule Predict The Charge Of An Ion Use The Electronegativity Values Of Atoms To Help Predict Whether An Ionic Or Covalent Bond Is Most Likely To Form Predict The Molecular Shape Of A Covalent Molecule Based Upon Its Lewis Dot Structure 4th, 2024

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Use The Bohr Models To Determine The Number Of Valance Electrons. Once You Have Found The Number Of Valance Electrons, Place Them Around The Elements Symbol. Element Atomic # Atomic Mass Protons Neutrons Electrons Valance Electrons Number Of Energy Levels Bohr Model Lewis Dot Carbon 6 12 6 6 6 4 (group 14, 4 In The Ones Palace) 2 (2nd Period) 3th, 2024

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Bohr Diagrams Directions: 1. Draw A Bohr Diagram For Each Element. 2. Write The Atomic Number In The Top Right Corner Of Each Box. 3. Write The Element Symbol Inside Of The Nucleus. 4. Add Energy Levels (shells) 3th, 2024

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