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1. Consider The Galvanic Cell Based On The Reaction $Al^{3+} + Mg \rightarrow Al + Mg^{2+}$ The Half Reactions Are $Al \rightarrow Al^{3+} + 3e^{-}$ $E^{\circ} = 1.66 V$ $Mg^{2+} + 2e^{-} \rightarrow Mg$ $E^{\circ} = 2.37 V$ Give The Balanced Cell Reaction And Calculate E° For The Cell. (See Work On Page 845) Solution: $0.71 V$
2. Using Jan 3th, 2024

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