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D. Uranium-235: 92 Protons, 143 Neutrons E. Carbon-12: 6 Protons, 6 Neutrons 3. Most Of An Atom's Mass Is Concentrated In The Nucleus. The Number Of Electrons And Protons Is The Same But Electrons Are So Light They Contribute Very Little Mass. The Mass Of The Proton Is 1,835 Times The Mass Of The Electron. Neutrons Have A Bit More Mass Than ... 3th, 2024

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(3) The Deuterium And Oxygen 18 Concentrations In Sea Water Usually Vary Parallely (FRIEDMAN, 1953).

(4) The Heavy Isotope Content In Precipitation Decreases With The Condensation Temperature, Which Is Reflected By (a) Variation Of The Composition Of Precipitation From Individual Atmospheric Cooling  
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