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Ideal Gas Law Worksheet PV = NRT

Ideal Gas Law Worksheet $PV = nRT$ Use The Ideal Gas Law, “ $PV = nRT$ ”, And The Universal Gas Constant $R = 0.0821 \text{ L}\cdot\text{atm} / (\text{K}\cdot\text{mol})$ To Solve The Following Problems: n = moles If Pressure Is Needed In kPa Then Convert By Multiplying By $101.3 \text{ kPa} / 1 \text{ atm}$ To Get $R = 8.31 \text{ kPa}\cdot\text{L} / (\text{K}\cdot\text{mole})$ Jan 15th, 2024

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A Perfect Radiator (“black Body”) Would Have $E=1$. A Perfect Reflector (“shiny” Object) Would Not Radiate At All; $E=0$. Lecture 33 10/28 Radiation This Behavior Is Contained In The Stefan-Boltzmann Law: Here, E Is Apr 9th, 2024

The Ideal And Combined Gas Laws PV = NRT Or P1V1 = P2V2 T ...

The Ascent, The Temperature Drops From 24°C To 5°C ? 11. An Unknown Gas Has A Volume Of 200 L At 5 atm And -140°C . What Is Its Volume At STP? 12. In An Autoclave, A Constant Amount Of Steam Is Generated At A Constant Volume. Under 1.00 atm Pressure The Steam Temperature Is 100°C . What Pre Jan 14th, 2024

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