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QΔt Where Δt = Difference Between Entering And Leaving Dry-bulb Temperature And Q = Airflow Rate In Cubic Feet Per Minute Latent Heat (Btu/h) = 0.68 QΔg Where Δg = Difference In Moisture Content Of Entering And Leaving Air, Grains Per Pound Of Dry Air Water Jan 1th, 2024

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23 0.0435 12 0.174 16 0.00726 0.003625 0.003625 0.003625 0.01088 0.003625 Empirical Formula = Na 2CO 3 2 1 3 5) 0.150 G Of Copper Reacts With Oxygen Form 0.188 G Of Copper Oxide. Find The Empirical Formula Of Copper Oxide. Mass Cu = 0.150 G Cu O Mass O = 0.188 – 0.150 = 0.038 G 0.150 63.5 0.038 16 0.00236 0.00236 0.002375 0.00236 Feb 3th, 2024

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