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Are Highly Toxic To The Environment, As Well As To

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Having Effective Technology To Treat These Metal
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Of PH On The Heavy Metal Ions Removal Using Ion Exchange Was Studied In The Range (5, 7, And 9) By Adding 0.1N HCL, 0.1N NaOH To Obtain The Desired PH Checked By A PH-meter. The Amount Of Clay Were 0.2, 0.5, 1, 1.5, And 2 Gm, The Con-centrations Of Heavy Metals Ions Were 10, 20, 30, 40, And 50, And Contact Times Were 15, 30, 45, Jan 4th, 2021

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Spectroscopy Were Used To Determine The Concentration Of These Ions.The Obtained Results Reveal That The Percent Adsorption For Removal Of Most Ions At (pH = 6.7, Particle Size Less Than 38 μm , And 3 Min Retention Time) Was Approximately 90%.
Keywords: Cellulose Acetate,heavy Metal, Column Tech, Ions Removal, Adsorption. 1. INTRODUCTION
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Used. The Mechanism Of Heavy Metal Removal By Precipitation Of Metal Hydroxides Is Shown In Eq. 1: $M^{2+} + 2OH^{-} \rightleftharpoons M(OH)_2$ Where M^{2+} And OH^{-} Represent The Dissolved Metal Ions And The Precipitant, Respectively, While $M(OH)_2$ Is The Insoluble Metal Hydroxide (Wang Et Al. 2004). For The Precipitation Of Heavy Metals From The Solu- Jan 2th,

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