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Chemistry E Cell = E Ind - E Ref (+ E J) Nernst Equation $E = E^0 - \frac{RT}{nF} \ln \frac{[Red]}{[Ox]}$ Where R = Gas Constant T = Absolute Temperature N = Number Of Electrons In Reaction F = Faraday's Constant E^0 = Standard Potential [Red] = Molar Concentration Of Reduced Form Of Speci Apr 1th, 2024 Chapter 5 Absorption And Stripping Which The Gas Flow Is Introduced In The Bottom Of The Column And The Liquid Solvent Is Introduced In The Top Of The Column. The Mathematical Analysis For Both The Packed And Plated Columns Is Very Similar. L X T A, t L X B A, b V Y T A, t V Y B A, b L V Figure 5.2-1 Countercurrent Absorption Process. The Overall Material Balance For A Countercurrent Absorption Process Is $L_b + V_t = L_t + V_b$ (5.2-1 ... Apr 2th, 2024.

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