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3) $2(aq) + C. K(s) + H_2SO_4(aq) + D. Zn(s) + NaOH(aq) +$

SECTION 11.3 REACTIONS IN AQUEOUS SOLUTION

1. Write The Net Ionic Equation For The Reaction Between Aqueous Barium Nitrate, $Ba(NO_3)_2$, And Sodium Sulfate, Na_2SO_4 .

2. Magnesium Reacts With HCl To Form Hydrogen And Magnesium Chloride. Write The Balanced Net Ionic Equation For This Reaction.

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11.1 Describing Chemical Reactions

11 Figure 11.1 The Separate Ingredi-ents Of The Bread Are Now Combined Into One Product—the Loaf Of Bread. Connecting To Your World Section 11.1 Describing Chemical Reactions 321 A B 11.1 Describing Chemical Reactions 2th, 2024.

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4. $\text{HCl(aq)} + \text{Sr(OH)}_2\text{(aq)} \rightarrow \text{SrCl}_2\text{(aq)} + \text{H}_2\text{O(l)}$
5. $\text{AlBr}_3\text{(s)} \rightarrow \text{Al(s)} + \text{Br}_2\text{(l)}$
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