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1.  $\underline{\quad}$  CaCl<sub>2</sub> +  $\underline{\quad}$  AgNO<sub>3</sub> →  $\underline{\quad}$  Ca(NO<sub>3</sub>)<sub>2</sub> +  $\underline{\quad}$  AgCl  
How Many Grams Of Silver Chloride Are Produced When 45 G Of Calcium Chloride React With Excess Silver ... 12th, 2024

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Stoichiometry: Mixed Problems (KEY) 1)  $N_2 + 3H_2 \rightarrow 2NH_3$  What Volume Of  $NH_3$  At STP Is Produced If 25.0 Of  $N_2$  Is Reacted With An Excess Of  $H_2$ ? 3 3 3 2 3 2 2 2 40.0L  $NH_3$  1mol  $NH_3$  22.4 9th, 2024

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### **Name: KEY**

Chemistry 11 Stoichiometry Review Worksheet

Name: \_\_\_KEY\_\_\_ 33.  $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$

Actual yield = 1.45g Theoretical yield = 5.95g  $\text{KClO}_4 \times 1\text{mol}$

$122.55\text{g} \times 2\text{O}_2 \times 32.00\text{g} \times 1\text{mol} = 3.11\text{g O}_2$

%Yield =  $1.45\text{g} \div 3.11\text{g} \times 100\% = 46.6\%$  34.  $4\text{FeCO}_3 + \text{O}_2$

$\rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$

Percent purity = 62.8%? pure  $\text{FeCO}_3$  19th, 2024

## **Chemistry Supplemental Stoichiometry Problems**

### **Answer Key**

Chemistry Supplemental Stoichiometry Problems

Answer Key Stoichiometry Is A Chemical Section

Involving The Use Of Relationships Between Reagents

And / Or Products In A Chemical Reaction To

Determine The Desired Quantitative Data. In Greek,

Stoikhein Means Element And Metricity Means

Measurement, Then Steichiometry Literally Translated



Means ... 2th, 2024

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