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## **Chemistry Stoichiometry Problem Sheet 2**

### **Answer Key**

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Materials Chemistry: Stoichiometry – Problem Sheet 1  
Directions: Solve Each Of The Following Problems.  
Show Your Work, Including Proper Units, To Earn Full Credit. 1. Silver And Nitric Acid React According To The Following Balanced Equation:  $3 \text{ Ag(s)} + 4 \text{ HNO}_3 \text{ (aq)} \rightarrow 3 \text{ AgNO}_3 \text{ (aq)} + 2 \text{ H}_2\text{O(l)} + \text{NO} \dots$  15th, 2024

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### **Answer**

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Stoichiometry Volume Problems Worksheet Answers  
Stoichiometry: Mixed Problems (KEY) 1)  $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$  What Volume Of  $\text{NH}_3$  At STP Is Produced If 25.0  
Of  $\text{N}_2$  Is Reacted With An Excess Of  $\text{H}_2$ ? 3 3 3 2 3 2 2 2

40.0L NH 1mol NH 22.4 15th, 2024

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## **Practice Work 53 Stoichiometry-04 Mixed Stoichiometry ...**

Practice Work 53 – Stoichiometry-04 Mixed Stoichiometry Problems General Information You Will Need A Periodic Table, Your Stoichiometry Notes, And Appendix 12 For This Assignment. Sorry About The Lack Of Format. I'm In A Time Crunch. 123.88 G/mol 70.90 G/mol 137.32 ... 12th, 2024

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Chemistry: Stoichiometry – Problem Sheet 1

Directions: Solve Each Of The Following Problems.

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Directions: Solve Each Of The Following Problems.

Show Your Work, Including Proper Units 1. \_\_\_ CaCl<sub>2</sub> + \_\_\_ AgNO<sub>3</sub> \_\_\_ Ca(NO<sub>3</sub>)<sub>2</sub> + \_\_\_ AgCl How Many Grams Of Silver Chloride Are Produced When 45 G Of Calcium Chloride React With Excess Silver ... 19th, 2024

## **A Practice Problem On Stoichiometry -- ANSWERS**

A Practice Problem On Stoichiometry -- ANSWERS

Consider The Reaction Represented By The Equation

Below For All Parts Of This Question:  $2 \text{ BrCl}_3 \rightarrow 3 \text{ Cl}_2 + \text{Br}_2$  (a) If 4 Moles Of  $\text{BrCl}_3$  Reacts According To The Equation, How Many Moles Of  $\text{Cl}_2$  Will Be Formed? How Many Moles Of  $\text{Br}_2$  Will Be Formed? Answers: 6 Mol  $\text{Cl}_2$  And 2 Mol  $\text{Br}_2$  Setups/Work: 4th, 2024

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