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Oxide Reacts With Br_2 And Gives Nitrosyl Bromide As Per Reaction Is Given Below:
 $2\text{NO}(\text{g}) + \text{Br}_2(\text{g}) \rightleftharpoons 2\text{NOBr} \dots$ Feb 5th, 2024CHEMISTRY (043) Class XI - CBSEClass XI
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