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Answer: The Scheme Below Represents Born-Haber

Cycle For BaO Crystal. Step 1. Sublimation Of Solid Ba:

$\text{Ba(s)} \rightarrow \text{Ba(g)}$, $\Delta H_1 = 180.0 \text{ KJ/mol}$ During This Process,

The Change In Enthalpy Is Positive, Energy Is Required

To Change The State Of Barium From Solid To Gaseous

(to Break Metal Lattice). Step 2. Mar 17th, 2024

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Sea Water Is A Major Source Of Iodine. The Iodine Extracted From Sea Water Is Impure. It Is Purified In A Two-stage Process. Stage 1 $\text{I}_2 + 2\text{H}_2\text{O} + \text{SO}_2 \rightarrow 2\text{HI} + \text{H}_2\text{SO}_4$ Stage 2 $2\text{HI} + \text{Cl}_2 \rightarrow 2\text{HCl} + \text{I}_2$ (a) (i) State The Initial Oxidation State And The Final Oxidation State Of Sulfur In Stage 1. [2 ... Feb 1th, 2024

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