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3 Another Example A Solution With Two Volatile Components •Modified Raoult's Law •VP T = X AP A + X BP B Example •At 20.0 OC The Vapor Pressures Of Methanol (CH 30H) and Ethanol (C 2H 50H) Are 95.0 And 45.0 MmHg Respectively. An Ideal Solution Contains 16.1 G Of Methanol Apr 11th, 2024

## Colligative Properties- Page 1 Lecture 4: Colligative ...

B Raises Boiling Point Of Water From 100 O C To 100.640 C •  $\Delta$ T F Lowers Freezing Point Of Water From 0o C To -2.320 C •  $\pi$  Raises Osmotic Pressure From 1 Atm To 28.6 Atm While The Temperature Was Changing By A Degre Jan 5th, 2024

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#### **WORKSHEET: SOLUTIONS AND COLLIGATIVE PROPERTIES SET A**

Answer: [K +] = 0.71 M, [SO42-] = 0.24 M, [PO4 3-] = 0.074 M. 4. Calculate The Concentration Of Each Ion And The Mass Of Any Precipitate When A 0.300 Mole Of Aluminum Hydroxide Is Added To 50.0 MI Of 2.5 M Nitric Acid Solution (Assume That There Is No Volume Change Upon Jan 11th, 2024

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5. What Is The Mass Of This Sample? ... Compare And Contrast Moment What Aspects Were The Same About Problem #3 And Problem #2? ... Calculate The Molecular Weight Of The Mysterious Compound. 1 Liter Of Solution H Mar 10th, 2024

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6. What Colligative Property Is Responsible For Antifreeze Keeping Water From Freezing In A Car's Cooling System? \_\_\_\_\_ 7. What Colligative Property Is Responsible For Antifreeze Keeping Water From Boiling Over During The Summer In A Car's Cooling System? \_\_\_\_\_ 8. What Colligative Property Causes Ice To Melt On The Street After Salt Has Been Apr 4th, 2024

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## 12.3 Colligative Properties - REMONDINI

Colligative Properties For Molecular Solute Vs. Ionic Solute. Example: 1.0 M Of Sugar & 1.0 M Of Salt Show Different  $\Delta T$  B And  $\Delta T$  F Even Though The Molality Concentrations Of The Solute Are Identical. Note That;  $\Delta T = M \cdot K$  F,b Where K Is Only A Function Of The Solvent !!! Reason For The Difference In Colligative Properties Must Be Based On ... May 17th, 2024

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