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4. $\text{HCl(aq)} + \text{Sr(OH)}_2\text{(aq)} \rightarrow \text{SrCl}_2\text{(aq)} + \text{H}_2\text{O(l)}$ 5. $\text{AlBr}_3\text{(s)} \rightarrow \text{Al(s)} + \text{Br}_2\text{(l)}$

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C. Formulas Of The Products. B. Subscripts. D. Number Of Products. ____ 24.
According To The Law Of Conservation Of Mass, The Total Mass Of The Reacting
Substances Is A. Always More Than The Total Mass Of The Products. B. Always Less
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