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Chemical Reactions And Quantities

Chemical Reaction • A Chemical Reaction Always Involves A Chemical Change Because Atoms Of The Reacting Substances Form New Combinations With New Properties. • For Example, A Chemical Reaction (and Chemical Change) Takes Place When A Piece Of Iron (Fe) Combine 16th, 2024

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Chemical Quantities

1-carbon 2-oxygen Thus One Mole Of CO₂ Contains Three Times Avogadro's # Of Atoms. Example: How Many Atoms Are In 2.12 Mole Of Propane (C₃H₈)? Know: # Of Moles = 2.12 Mole C₃H₈ 1 Mole C₃H₈ = 6.02 X 10²³ Molecules C₃H₈ 1 Molecule C₃H₈ = 11 Atoms (3 Carbon Atoms & 8 Hydrogen Atoms) 15th, 2024

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NOTES: 10.1-10.2 Chemical Quantities (The Mole / Molar ...

Chemical Quantities -consider Another Word That Means A #: Measuring Doughnuts: A DOZEN! 1 Dozen = 12 Doughnuts (count) 1 Dozen = 500 G Doughnuts (mass) 1 Dozen = 1 Box Doughnuts (volume) Measuring Hydrogen (H₂ Gas): A MOLE! 1 Mole = 6.02 X 10²³ H₂ Molecules (count) 1 Mo 1th, 2024

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