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$H_0 = 92 \text{ KJ Mol}^{-1}$ $S_0 = 192 \text{ J Mol}^{-1} \text{ K}^{-1}$ 1th, 2024
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Exam 1 15% Exam 2 15% Exam 3 15% Final Exam
20% Research Paper 10% Homework 20% Class
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L.35 In The Textbook Question: In The Third Reaction Step The Compound "FeBr₂" Needs To Be "Fe₃Br₈".

1A.15 Typo. Answer Is $N = 3$ To $N = 1$ As Energy Is Emitted. 1B.27 Solution Manual States $\Delta v = 5.0 \text{ M.s}^{-1}$ Velocity Was Given As $5.00 \pm 5.0 \text{ M.s}^{-1}$ 1th,

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