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Thermochemistry © 2009, Prentice-Hall, Inc. Energy •Energy Is The Ability To Do Work Or Transfer Heat. -En May 3th, 2024

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162 CHAPTER 6: THERMOCHEMISTRY To Convert The Answer To Joules, We Write: $101.3 \text{ J} \cdot 0.18 \text{ L} \cdot \text{Atm} = 18.2 \text{ L} \cdot \text{Atm} = 18.2 \cdot 101.3 \text{ J} = 1843 \text{ J}$ 6.17 An Expansion Implies An Increase In Volume, Therefore W Must Be -325 J (see The Defining Equation For Pressure-volume Work.) If The System Absorbs Heat, Q Must Be $+127 \text{ J}$. The Change In Energy (internal Mar 3th, 2024

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