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#### **Chapter 3 Stoichiometry**

Chapter 3 Stoichiometry 3-7 EXAMPLE PROBLEM: Determine The Molar Mass Of A Compound Calculate The Molar Mass For Each Of The Following Compounds: (a) 2-Propanol,  $\text{CH}_3\text{CH}(\text{OH})\text{CH}_3$  (b) Iron(II) Phosphate,  $\text{Fe}_3(\text{PO}_4)_2$  SOLUTION: You Are Asked To Calculate The Molar Mass Of A Compound. You Are Given The Compound's Formula. 5th, 2024

#### **CHAPTER 2: Atoms, Molecules And Stoichiometry**

2.5 Stoichiometry And Equations Stoichiometry 1) Stoichiometry Is The Proportion Of Things Either Reacting Or Combining. 2) In Compounds, It Refers To The Ratio In Which The Atoms Are Combined Together. For Example, Water,  $\text{H}_2\text{O}$  Has A Stoichiometry Of 2 Hydrogen To 1 Oxygen. 3) It Also Refers To The Reacting Proportions In A Chemical Equation. 2th, 2024

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