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States That The Product Of The Pressure And Volume Remains Constant As Long As n And T Are Unchanged. Assign The Symbols P_1 And V_1 To The Pressure And Volume Of The Tank And The Symbol P_2 To The Pressure Of A Balloon. According To Boyle's Law, The Volume Of A Gas Varies Inversely With Its Pressure If Temperature Is Constant.

Boyle's Law Problems From Book Solving Volume-Temperature Problems
Example: A gas is immersed in a container that remains constant. Gather information. The initial volume and temperature and the final temperature are given. In gas law problems, always convert temperatures to Kelvins. The gas law equations do not work for temperature in Celsius.
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