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November 22, 2017 2:09 PM Stoichiometry Feb 14th, 2024.

Basic Stoichiometry Practice Problems Answer Key  
Basic Stoichiometry Practice Problems Answer Key  
1) How Many Moles Of Ammonium Sulfate Can Be Made From The Reaction Of 30.0 Mol Of  $\text{NH}_3$  With  $\text{H}_2\text{SO}_4$  According To The Following Equation:  
 $\text{NH}_3 + \text{H}_2\text{SO}_4 \rightarrow (\text{NH}_4)_2\text{SO}_4$   
15.0 Mol  $\text{NH}_3$   
2) In A Very Violent Reaction Called A Thermite Reaction, Alumi Jan 29th, 2024

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