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$$CcO_2 = (Hb \times 1.34 \times SaO_2) + (PAO_2 \times 0.003)$$

Mixed Venous Oxygen Content (CvO_2)

$$CvO_2 = (Hb \times 1.34 \times SvO_2) + (PvO_2 \times 0.003)$$

Shunt Equation (Q_S/Q_T)

$$Q_S/Q_T = (CcO_2 - CaO_2) / (CcO_2 - CvO_2)$$

Modified Shunt Equation (Q_S/Q_T)

$$Q_S/Q_T = ((PAO_2 - PaO_2) \times 0.003) / ((CaO_2 - CvO_2) + (PAO_2 - PaO_2) \times 0.003)$$

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Percentage Composition Example: A Sample Of A Liquid With A Mass Of 8.657 G Was ... •Smallest Is 3.722×10^{-3} Mole •C = •H = Step 3: Divide All Number Of Moles By The Smallest 19th, 2024Chapter 3 Calculations With Chemical Formulas And EquationsChapter 3 Calculations With Chemical Formulas And Equations. Copyright © Cengage Learning. All Right 22th, 2024.

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C THREE CALCULATIONS WITH CHEMICAL FORMULAS ...Part Two: Determining
Chemical Formulas A. Percent Composition From Compound Formulas. (Section 3.3)

1. If Formula Of Compound Is Known, Its Percent Composition By Mass Of Each
Element Can Be Found, And Vice Versa. \div Atomic Masses (AM) Compound Percent
Composition Formula In Terms Of Elements X Atomic Masses (AM) 2. 26th, 2024

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