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In Their Water Of Crystallization. When They Are Heated Past The Melting Point And  
Subsequently Allowed To Cool, The Aqueous Solution Becomes Supersaturated. This  
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67 PREPARATION AND TESTING OF BUFFER SOLUTIONS. P. URPOSE. The Purpose Of The Laboratory Investigation Is To Experimentally Determine (1)  $pK_a$  (and Thus  $K_a$ ) Of The Acid In A Buffer And Thus The Buffer Range, (2) Investigate The Buffer Capacity Of May 6th, 2024

### **Section 19.1. Acid-Base Buffer Solutions**

If We Add  $\text{CH}_3\text{COO}^-$ -SHIFT Net Effect:  $[\text{H}^+]$  Decreases pH Increases Also,  $[\text{OH}^-]$  Increases  $\text{POH}$  Decreases I.e., a Weak Acid And Its Conjugate Base Or A Weak Base And Its Conjugate Acid. 19-2 Example: Addition Of  $\text{NH}_4\text{Cl}$  To  $\text{NH}_3$  May 2th, 2024

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