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Decomposition
Decomposition Reaction- One Substance Breaks Down, Or Decomposes, Into Two Or More Substances. $AB \rightarrow A + B$ Most Decomposition Reactions Require The Use Of Heat, Light, Or Electricity. May 4th, 2024

Classifying Chemical Reactions
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Section 8-3: Classifying Chemical Reactions
Section 8-3: Classifying Chemical Reactions
Reaction Type General Format Characteristics Of Reactants Characteristics Of Products Example (s) Decomposition $AB \rightarrow A + B$ Only One Reactant Two Or More Simpler Products $CuCO_3 (s) \rightarrow CuO + CO_2 (g)$ $2 H_2 O (l) \rightarrow 2 H_2 (g) + O_2 (g)$ Synthesis AKA: Combination $O (l) + B \rightarrow AB$ 2 Two Elements Or Two Simple May 6th, 2024.

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