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#82 Notes Unit 10: Acids & Bases Ch. Acids, Bases, And Solubility Ch. Acids/ Bases ** Acids: Start With "H" Or End In $-CO_2H$ ($-COOH$) And Have A Sour Taste. ** Bases: End In $-OH$ (positive Ion In Front) Or End In $-NH_2$, $-NH$. 2, NH_3 . 3. They Have A Bitter Taste And Feel Slippery. * 3 Acid/Base Theories . I. Arrhenius Theory A) Acids Produce H^+ Feb 4th, 2024

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Chapter 19 Acids, Bases, And Salts209 SECTION 19.1 ACID-BASE THEORIES(pages 587-593) This Section Compares And Contrasts Acids And Bases As Defined By The Theories Of Arrhenius, Brønsted-Lowry, And Lewis. It Also Identifies Conjugate Acid-base Pairs In Acid-base Reactions. Chapter May 2th, 2024

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Acids And Bases Can Also Be Identified Using An Operational Definition. Operational Definitions Are Simply A List Of Properties. ACIDS: ♦ A _____ Taste Is A Characteristic Property Of All Acids In Aqueous Solution. ♦ Acids React With Some Metals To Produce _____ Gas. ♦ Because Mar 7th, 2024

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