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TABLE OF CONJUGATE ACID-BASE PAIRS Acid Base K_a (25 C) TABLE OF CONJUGATE
ACID-BASE PAIRS Acid Base K_a (25 OC) HClO₄ ClO₄⁻ - H₂SO₄ HSO₄⁻ - HCl Cl⁻
HNO₃ NO₃⁻ - H₃O⁺ + H₂O H₂CrO₄ HCrO₄⁻ - 1.8 X 10⁻¹ H₂C₂O₄ (oxalic Acid)
Mar 2th, 2024 Chapter 3 Acid-Base Equilibria Acid Base Equilibria ... Chapter 3 Acid-
Base Equilibria Acid-Base Equilibria Acids And Bases Play A Key Role In A Number Of
Environmentally Important Chemical Reactions, Including Weathering, Transport Of
Metals In Solution, And CO₂ Atmosphere-water Equilibria. In This Chapter We Will
Develop The Concept Of An Acid And A Base, Characterize Strong And Weak Acids,
May 3th, 2024 (aq) Acid Base Conjugate Conjugate Acid Base Acid Base Conjugate

Conjugate . Acid Base . 2) What Is The Strongest Base In The Following Reaction?
 $\text{HNO}_3(\text{aq}) + \text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{NO}_3^-(\text{aq}) + \text{H}_3\text{O}^+(\text{aq})$ H_2O Is The Strongest Base. Strong
Acids, Such As HNO_3 Have Weak Conjugate Bases, So NO_3^- is A Weak Base. H 2O
And May 5th, 2024.

SECTION 19.1 ACID BASE THEORIES (pages 587-593) 1 Name: Period: Chapter 19
Acids, Bases, And Salts Final Exam Study Guide SECTION 19.1 ACID-BASE THEORIES
(pages 587-593) Define The Following Vocabulary Terms: 1. Monoprotic Acids 2.
Diprotic Acids 3. Triprotic Acids 4. Hydronium Ion (H_3O^+) 5. Amphoteric Properties Of
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